

Déposition de molécules organiques par électrospray

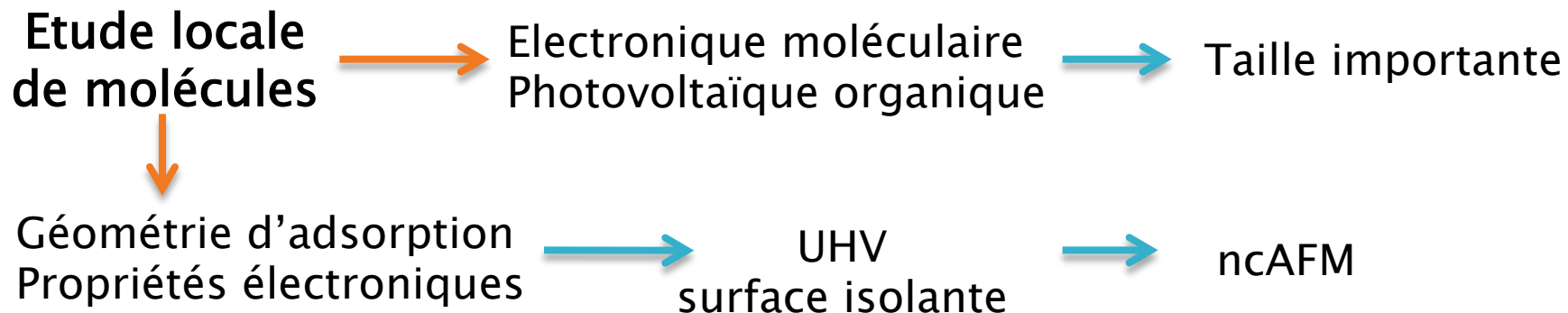
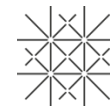
Hinaut Antoine, Rémy Pawlak, Thilo Glatzel et Ernst Meyer

Department of Physics, University of Basel, CH-4056 Basel

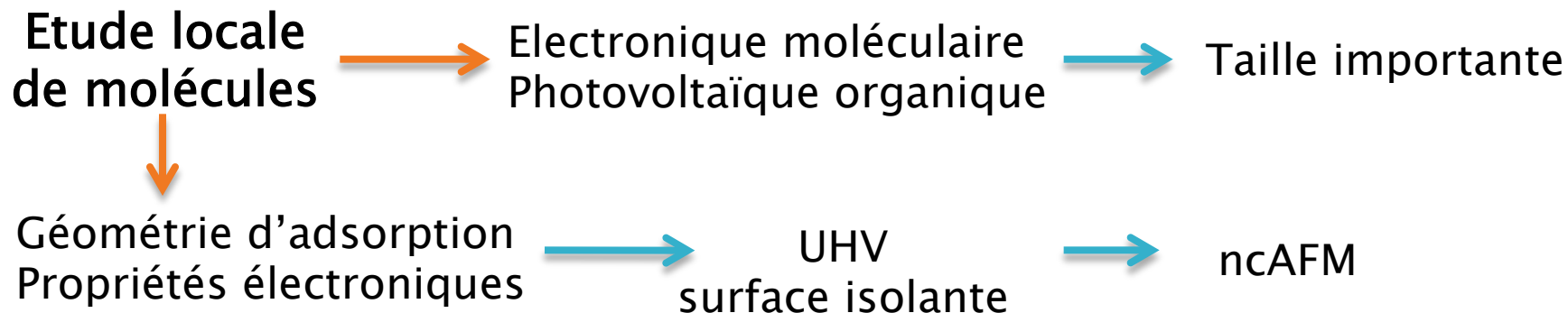
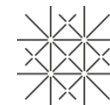
<https://nanolino.unibas.ch>



Introduction



Introduction



- ◆ Evaporation thermique : Technique la plus utilisée
T° sublimation vs T° fragmentation
- ◆ Plusieurs solutions :
Vanne pulsée, Droplet Casting ou Electrospray par Ionisation (ESI).

Tanaka et al., Surface Science (1999)

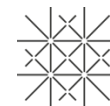
Zambelli et al., Int. J of Nanoscience (2004)

ESI pour déposer des molécules.

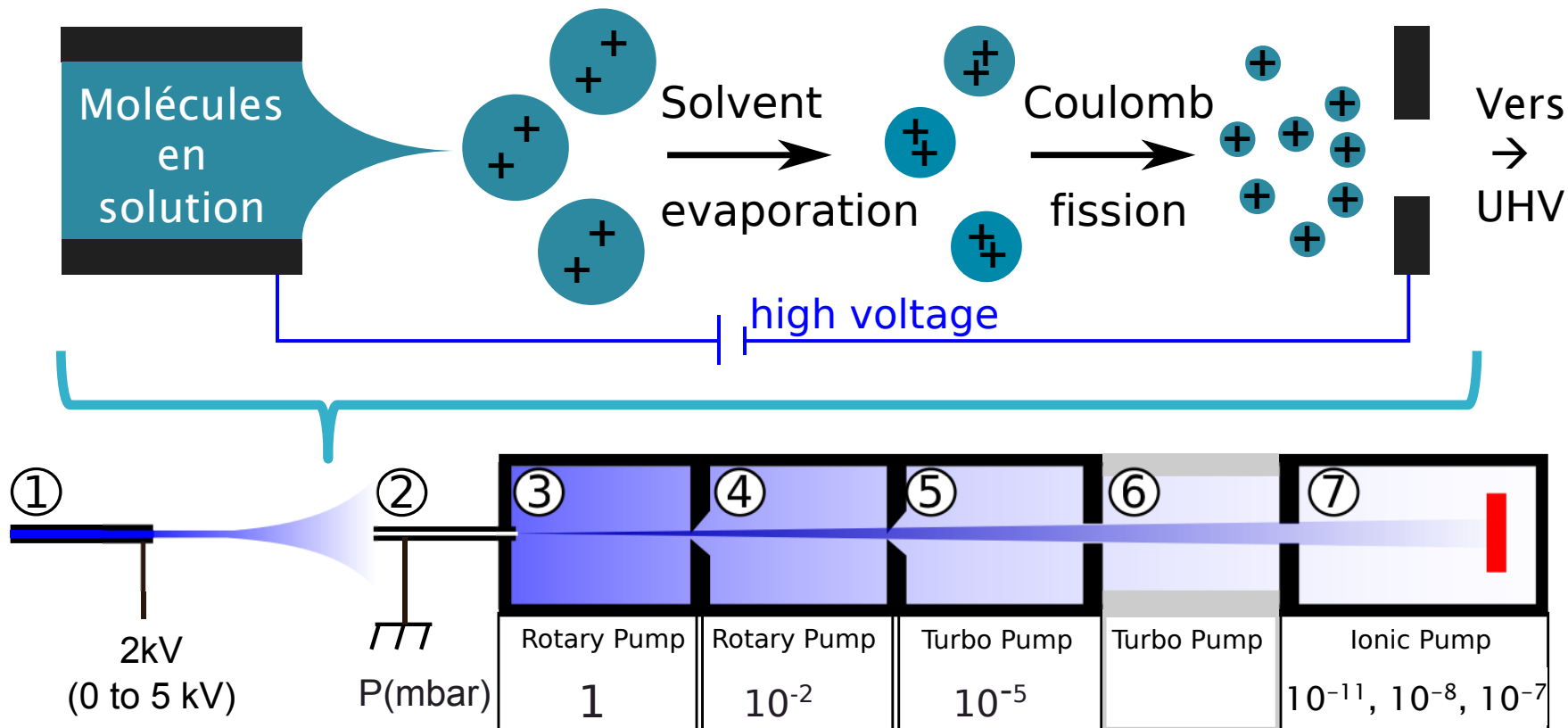
↓
Introduction de protéines sous vide pour leur étude par spectroscopie de masse.

J.B Fenn et al. (1989), Science 246

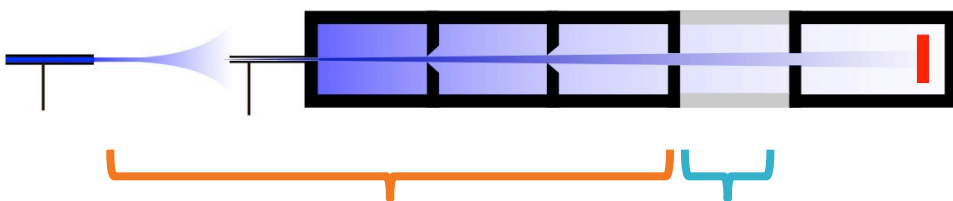
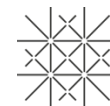
ElectroSpray Ionisation (ESI)



- Technique non destructive pour l'introduction sous UHV.
- Orienté vers un échantillon pour le dépôt de molécules.



Dispositifs :



Dispositifs avec sélection et tri des ions, soft landing :

S. Rauschenbach et al. (2006), Small, vol. 2.

C. Hamann et al. (2011), RSI, vol. 82.

A. Bodin et al. (2013), RSI, vol. 84.

Molecularspray :

Saywell et al., Angewandte Chemie 49 (2010)

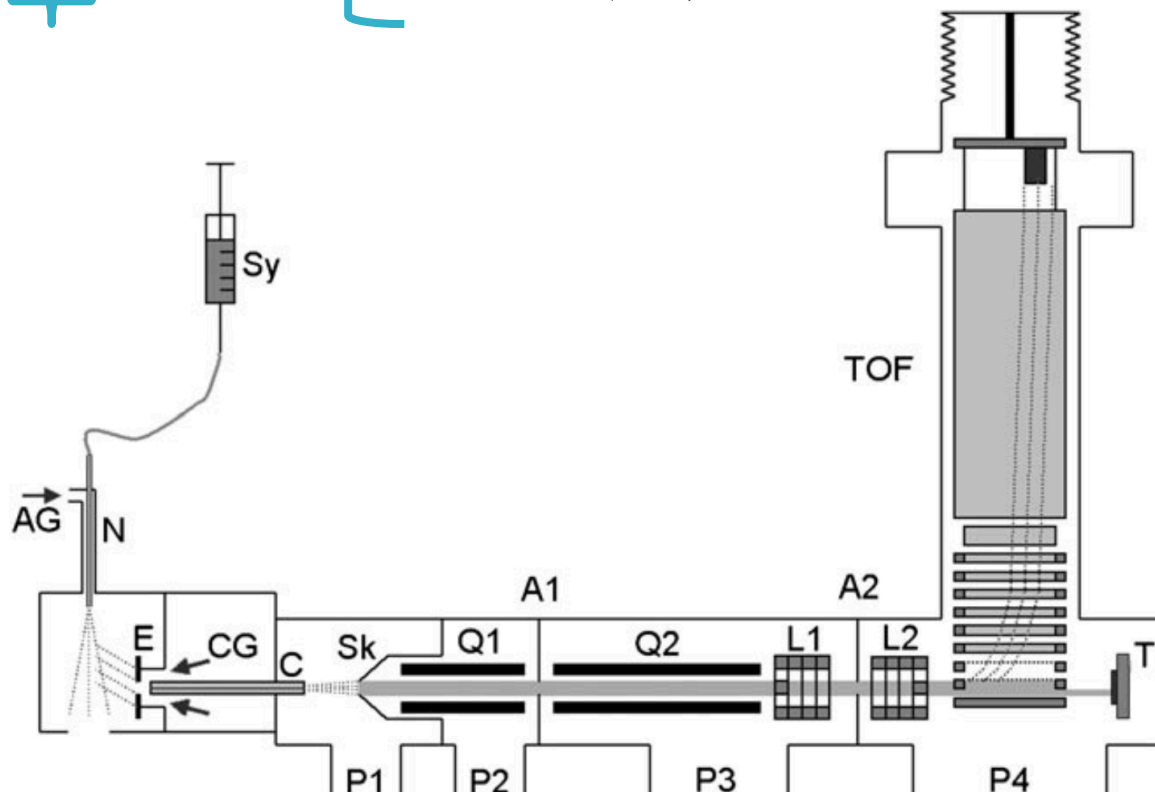
ADCAP Vacuum Technology :

Yokoyama et al., JPCC 117, 36 (2013)

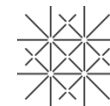
Nouvel étage de pompage :

A. Hinaut et al., BJ Nano 6 (2015).

P. Erler et al., ACS Nano 15 (2015).



Molecularspray



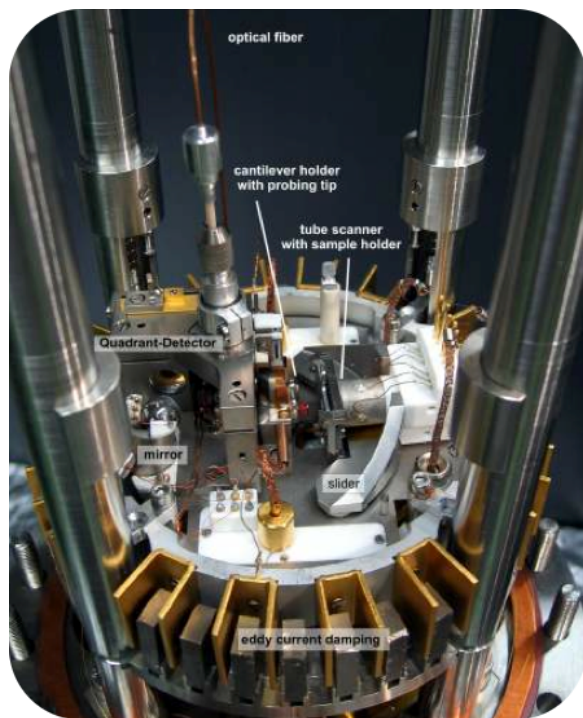
Chambre d'analyse
ncafm

10^{-11} mbar

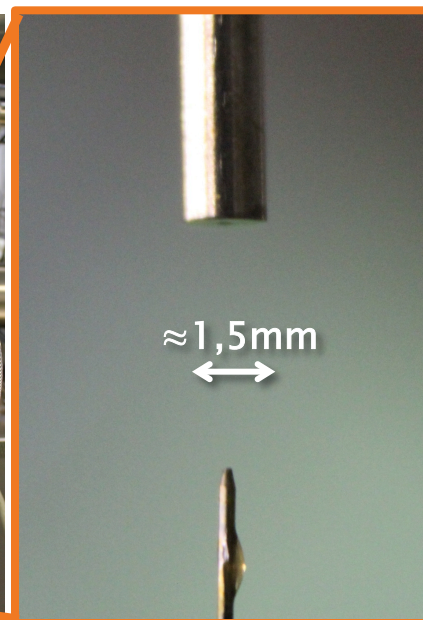
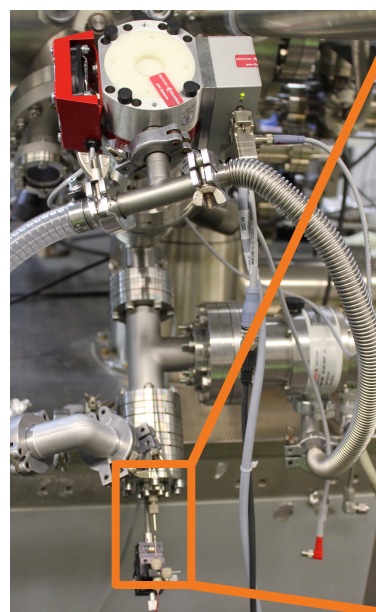
Chambre de
préparation

10^{-10} mbar

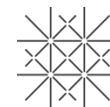
Sas d'introduction
rapide



ncAFM room
temperature,
home built



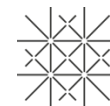
Spray



≈ 1,5mm

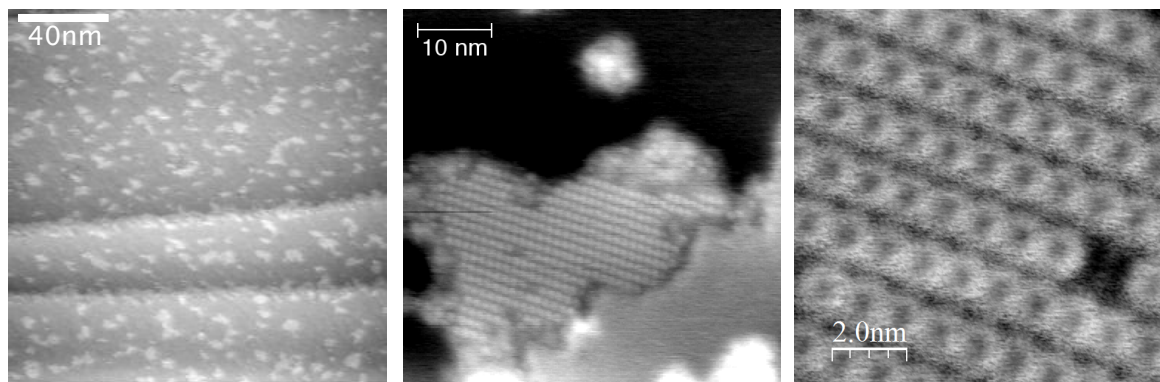
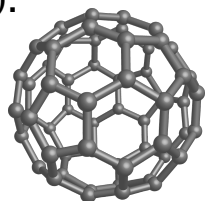


Dispositif : MolecularSpray



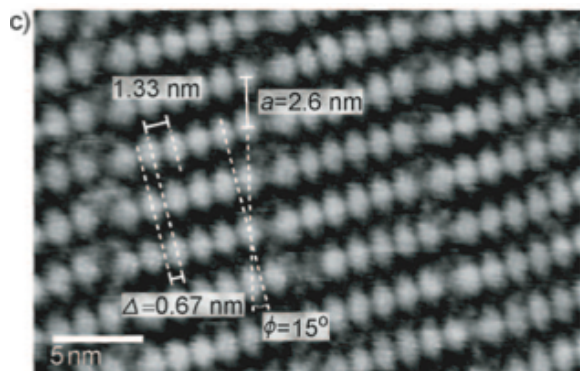
C₆₀/Cu(111)

Spray deposition = 3min.
Dichloromethane et
Acetonytrile (4:1).

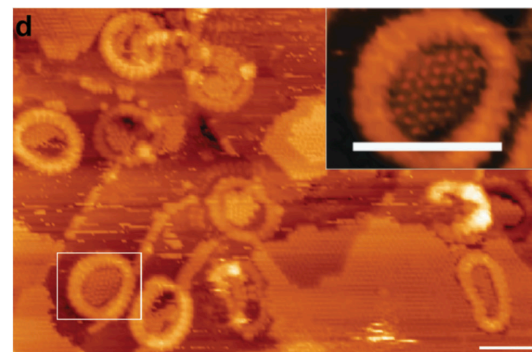
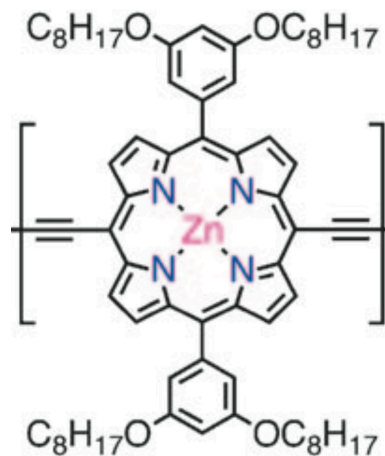


150°C 1h

**Porphyrin chain/Au(111),
chauffage 100°C.**

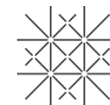


*A. Saywell et al., Angewandte
chemie, 49 (2010)*

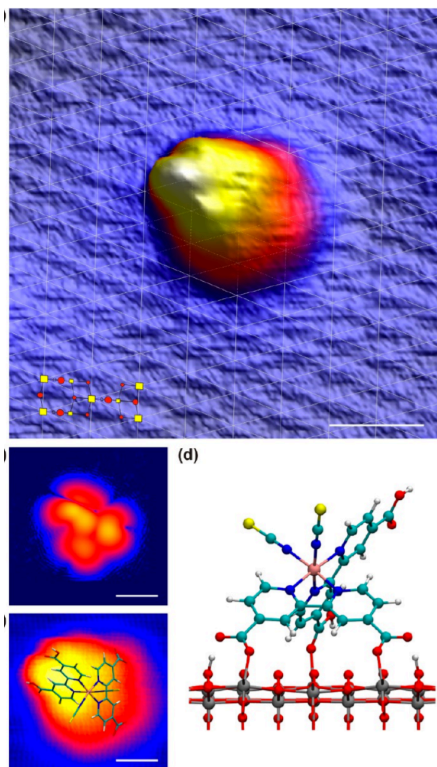


*M.B. Wieland et al., Chem.
Commun., 50 (2014)*

ESD avec sélection :

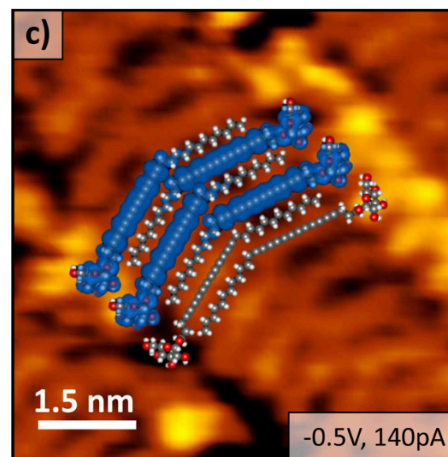
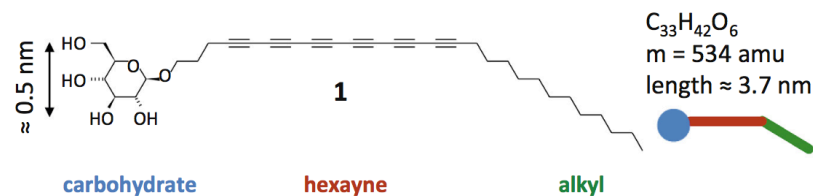


N3 Dye/TiO₂ anatase



*Kley C. S. et al., Nano Letters 14 (2014)
Groupe Klaus Kern*

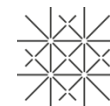
Glycosylated hexayne/Au(111)



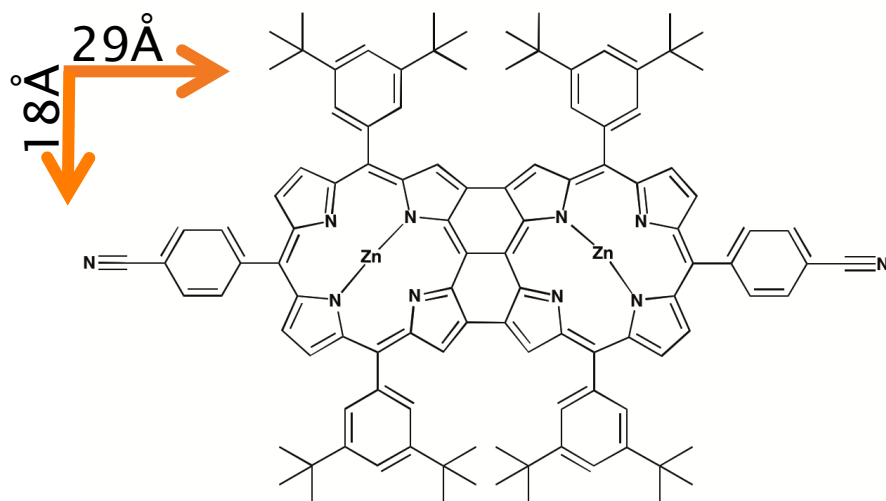
Quadrupole
&
Soft landing

*Rinke G. et al., International Journal of Mass Spectrometry (2015)
Groupe Richard Berndt*

Molécules et surface



Double porphyrine



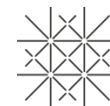
Synthèse par F. Diederich, ETH Zurich.
2 groupes CN : ancrage sur la surface

B. Such et al. (2010), ACS Nano 4.

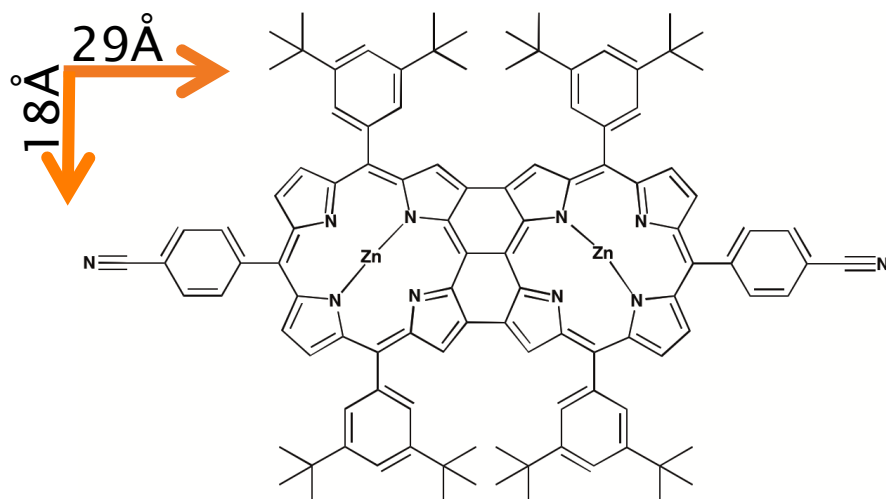
A. Hinaut et al. (2012), Beilstein J Nanotechnol 3.

Solvants type chromasolv
Toluène et isopropanol (2:1)

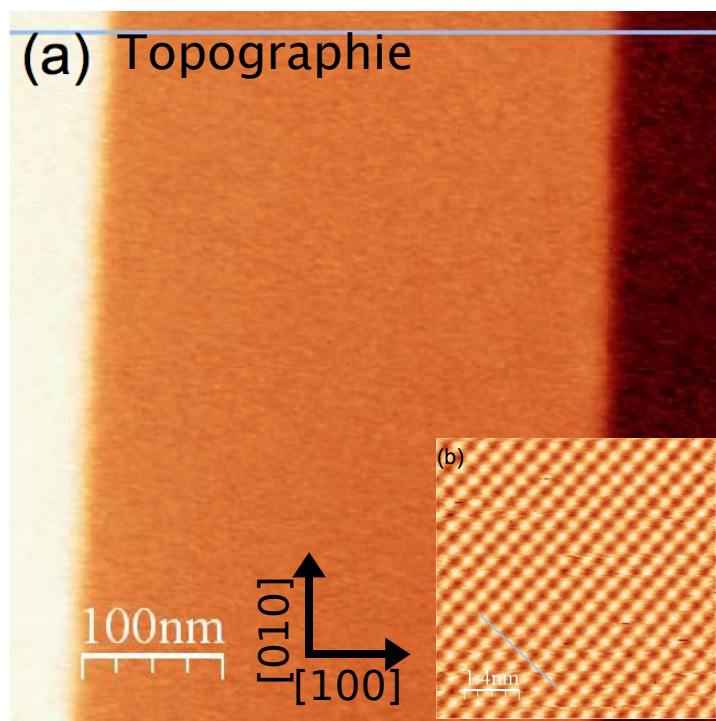
Molécules et surface



Double porphyrine



KBr(001)



Clivage air ou UHV.
→ Chauffage 1h @ 150°C.
→ -1V < Bias < 1V

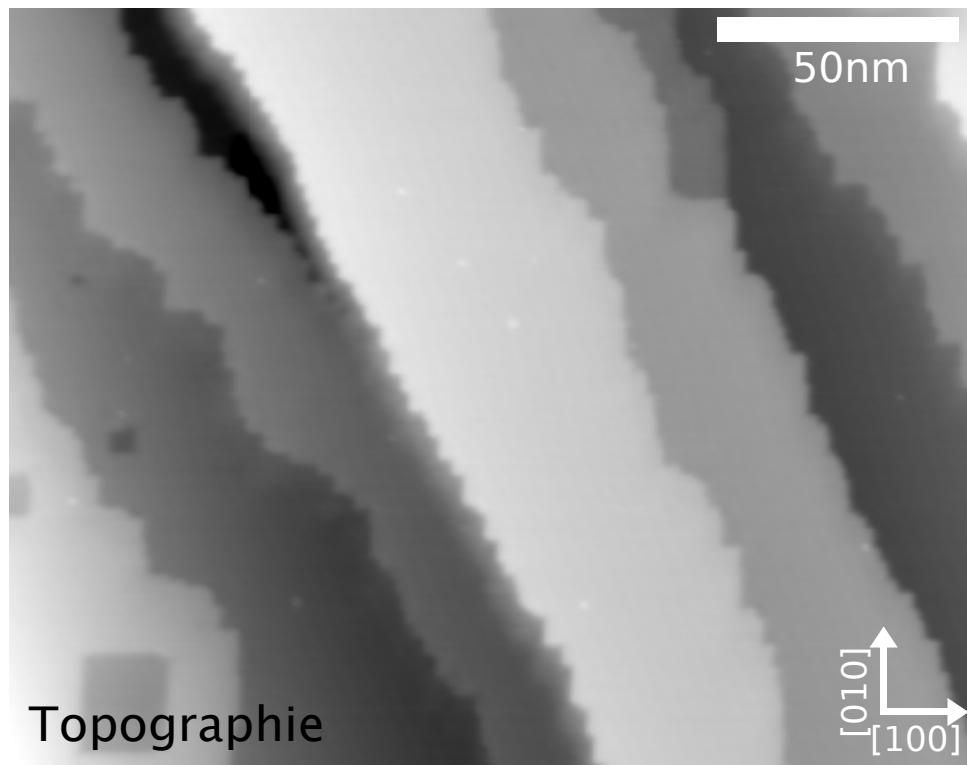
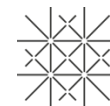
Synthèse par F. Diederich, ETH Zurich.
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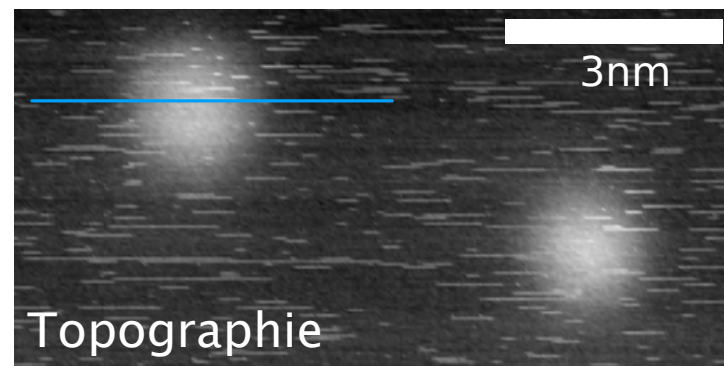
A. Hinaut et al. (2012), Beilstein J Nanotechnol 3.

Solvants type chromasolv
Toluène et isopropanol (2:1)

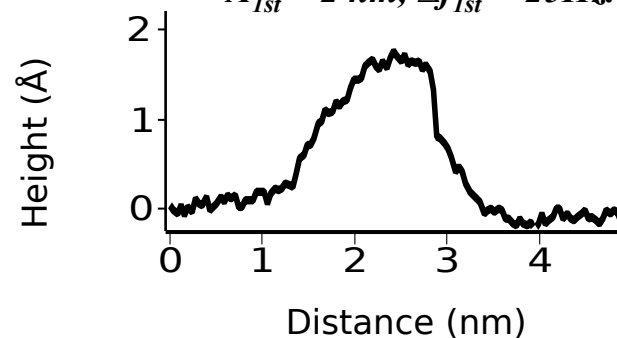
ESI : Solvant seul



$$A_{1st} = 5 \text{ nm}, \Delta f_{1st} = -10 \text{ Hz}$$

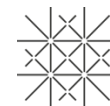


$$A_{1st} = 2 \text{ nm}, \Delta f_{1st} = -25 \text{ Hz}$$

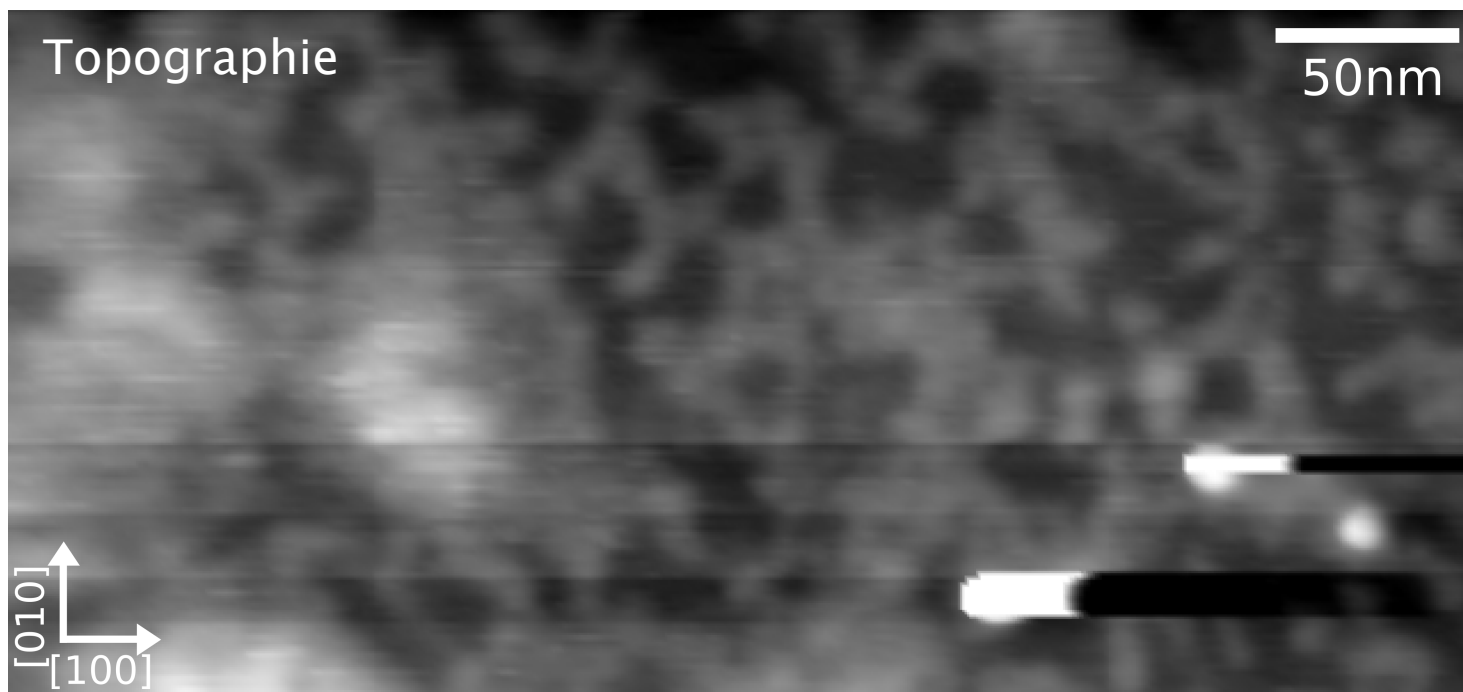


ESI : Toluène et isopropanol (2:1) pendant 30 min
 $1 \text{ kV} < U < 2,5 \text{ kV}$

ESI : Double porphyrine



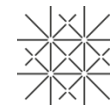
ESI : Double porphyrines dans toluène et isopropanol (2:1) pendant 5 min



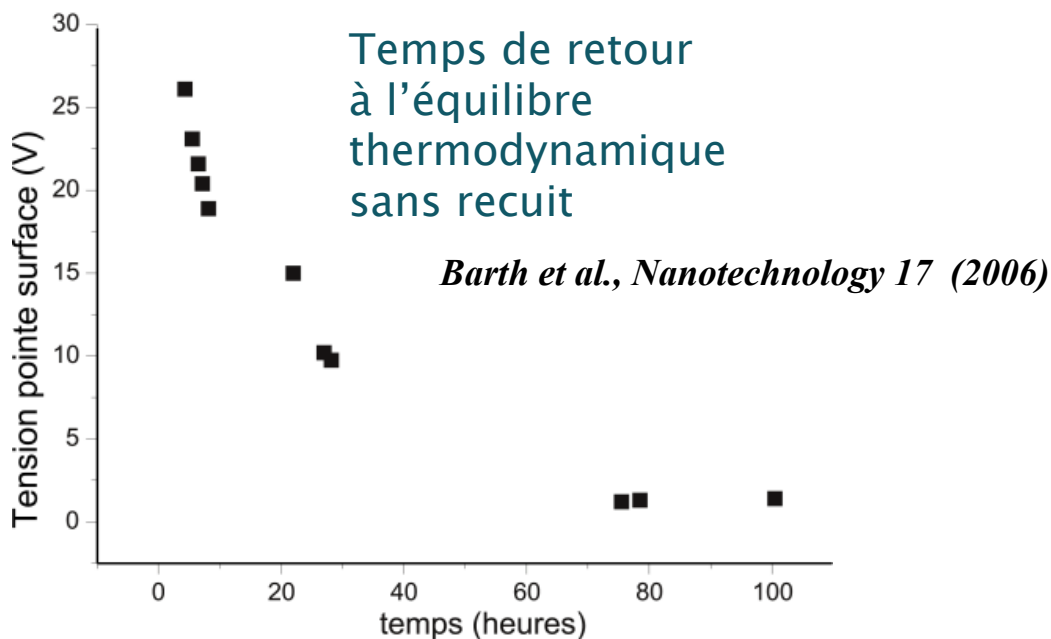
$A_{1st}=2nm, \Delta f_{1st}=-15Hz$

$U < -30V$ (-10V appliqués).

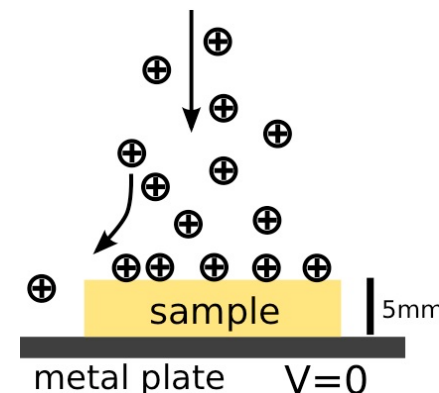
Cristaux ioniques et charges



Clivage sans recuit



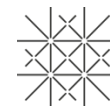
ESD sur isolant bulk :



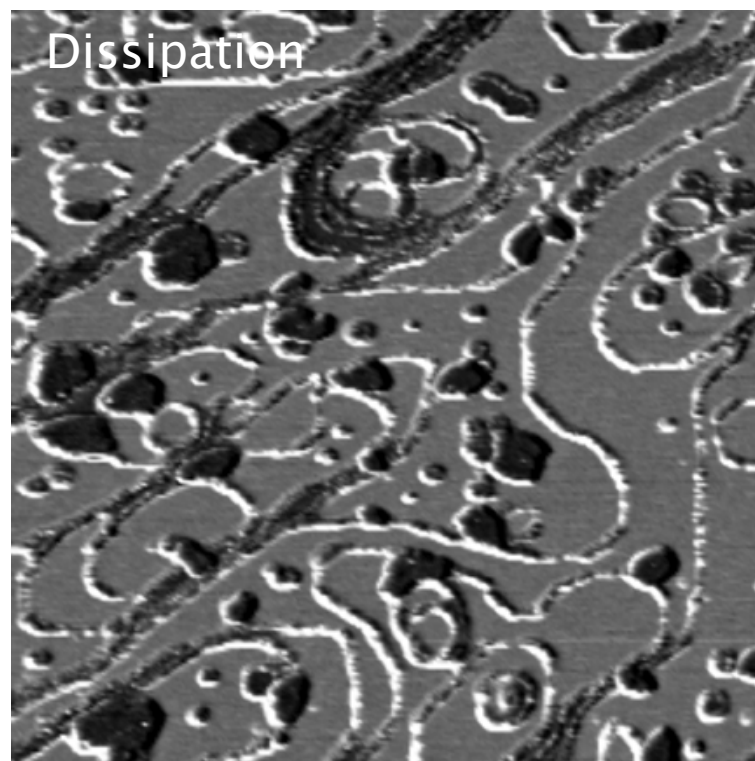
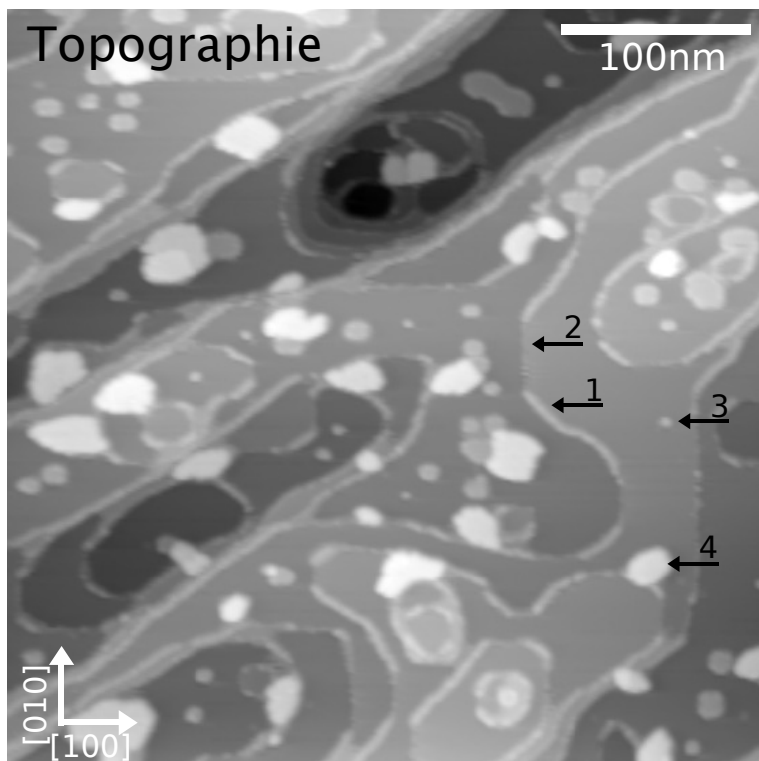
Répulsion électrostatique
→ taux de couverture limité.

→ Chauffage de l'échantillon Double porphyrine/KBr(001)
1h @ 350K

Double porphyrine : Recuit



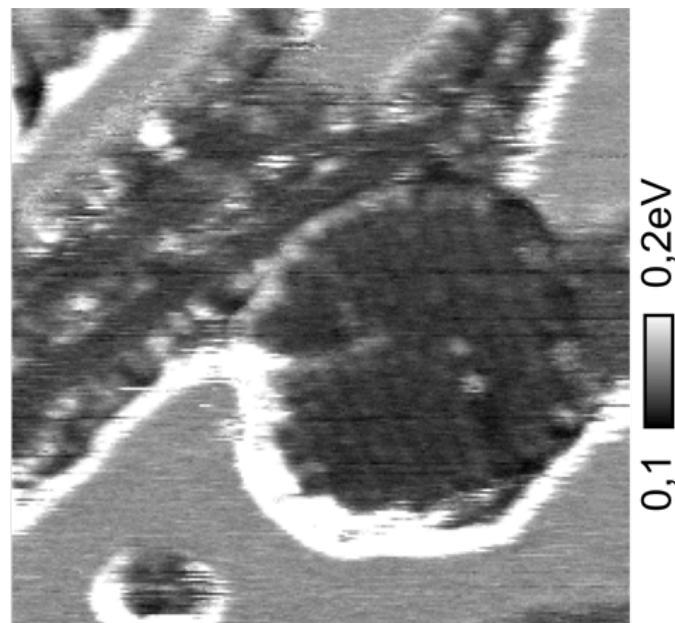
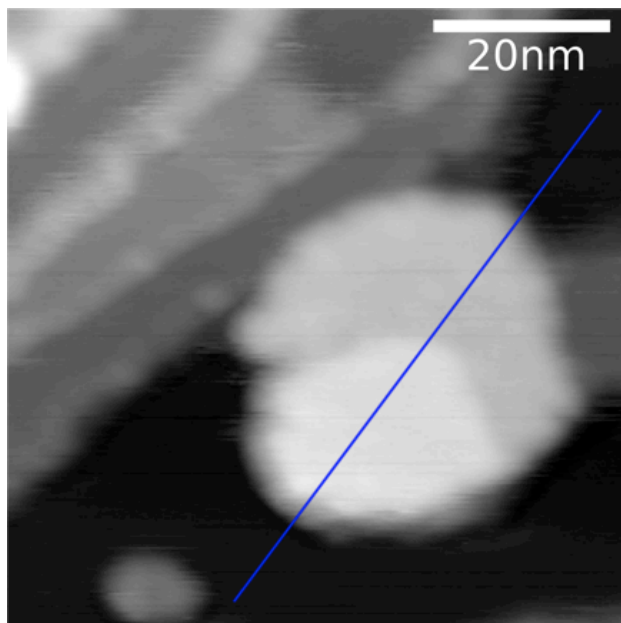
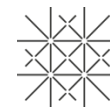
Chauffage échantillon : 1h @ 350K.



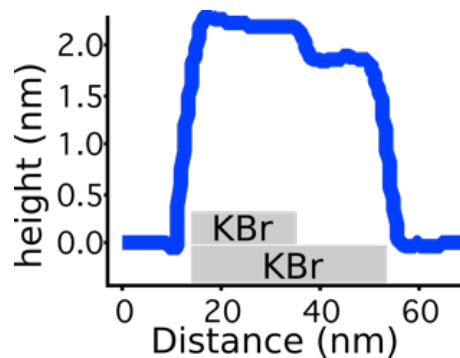
$A_{1st} = 4 \text{ nm}$, $\Delta f_{1st} = -10 \text{ Hz}$
 $U = -1V$

- 1 : bord de marche avec molécules adsorbées.
- 2 : bord de marche seul.
- 3 : petits agrégats.
- 4 : îlots.

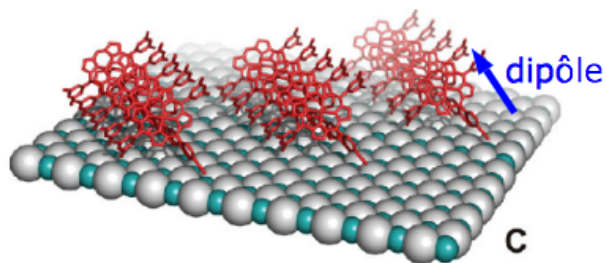
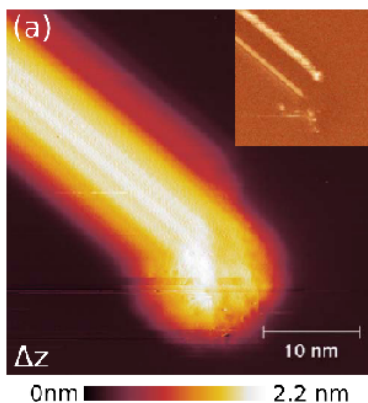
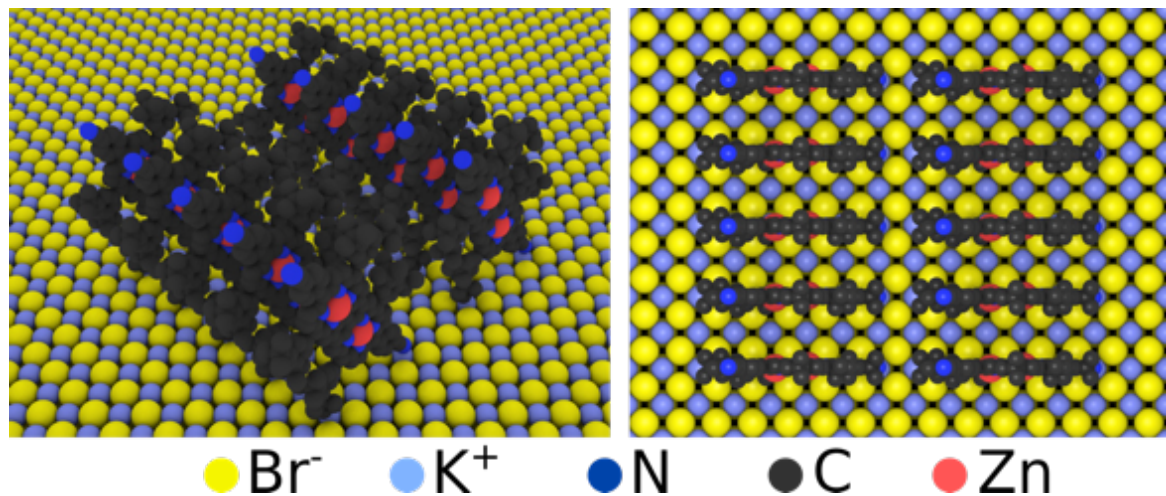
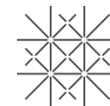
Double porphyrine : Recuit



$$A_{1st} = 5 \text{ nm}, \Delta f_{1st} = -8 \text{ Hz}$$

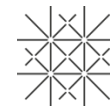


Double porphyrine : Recuit



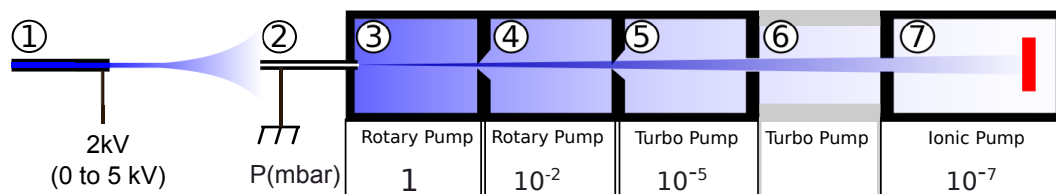
T. Glatzel et al. (2009), Applied Physics Letters 94.
A. Hinaut et al. (2012), Beilstein J Nanotechnol 3.

Double Porphyrine : Molécules uniques

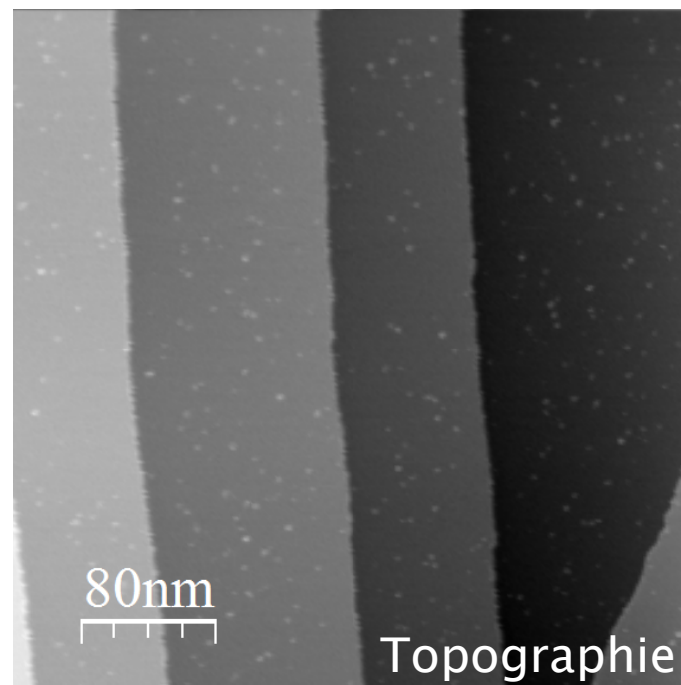


Modification du dispositif MolecularSpray.

- ↳ Ajout d'un nouvel étage de pompage différentiel.
- ↳ Eviter recuit des échantillons.

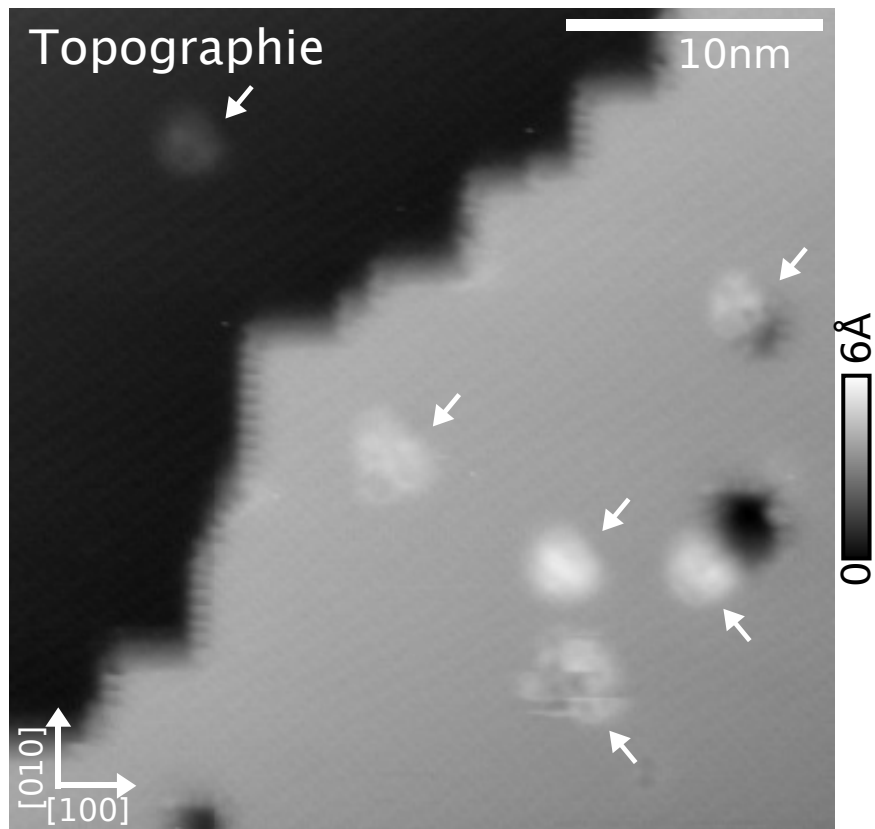
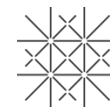


Double porphyrine / KBr(001)
Toluène et isopropanol (2:1)
 $U < 1,5\text{kV}$
5 min

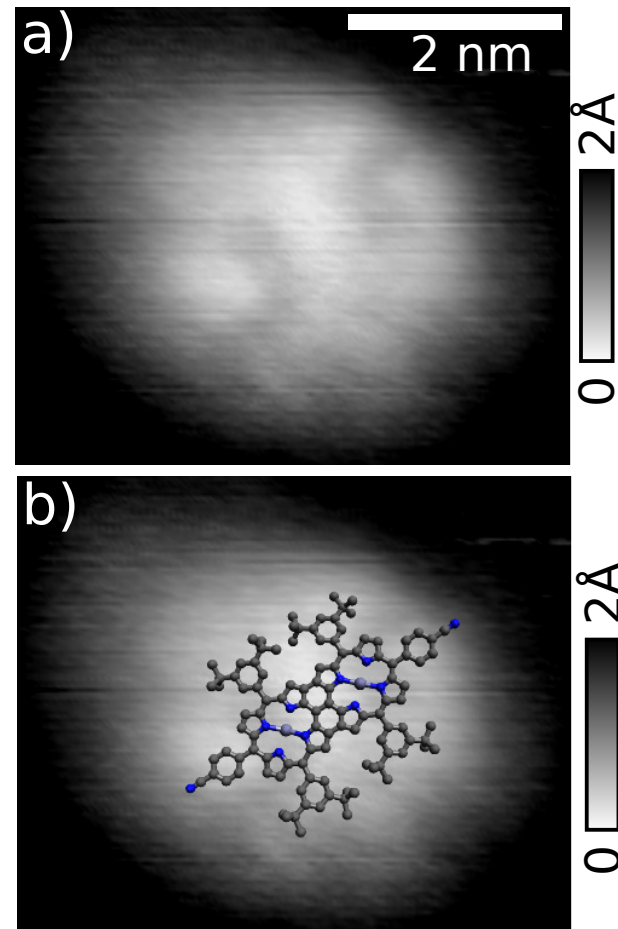


$f_2 = 1017\text{kHz}$, $A_2 = 400\text{pm}$, $\Delta f_2 = -10\text{Hz}$.

Double Porphyrine : Molécules uniques

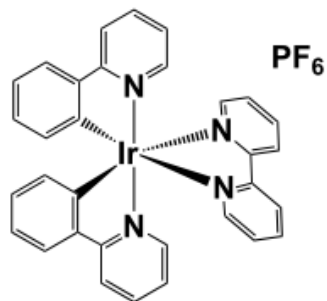
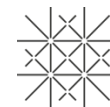


$$f_{2nd} = 1.02 \text{ MHz}, A_{2nd} = 400 \text{ pm}, \Delta f_{2nd} = -57 \text{ Hz}$$

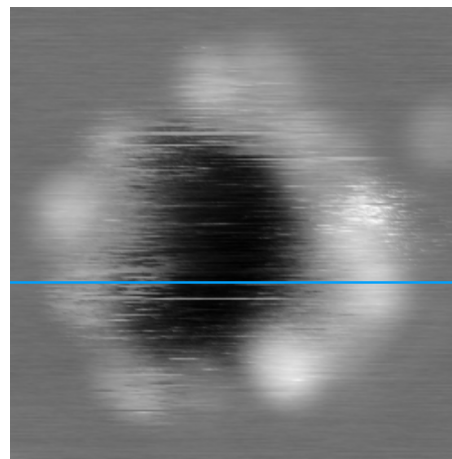
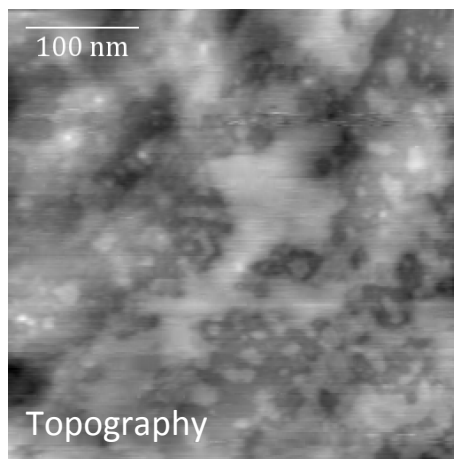


$$f_{2nd} = 1.02 \text{ MHz}, A_{2nd} = 400 \text{ pm}, \Delta f_{2nd} = -70 \text{ Hz}$$

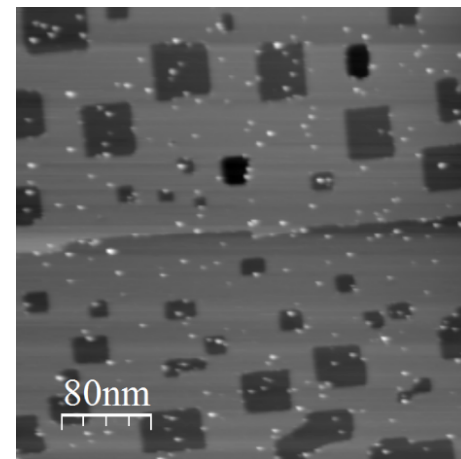
Ir Dye / KBr(001)



dichloromethane
et methanol (1:1)



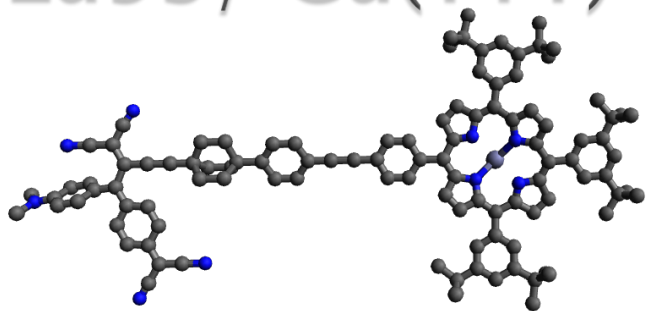
after annealing



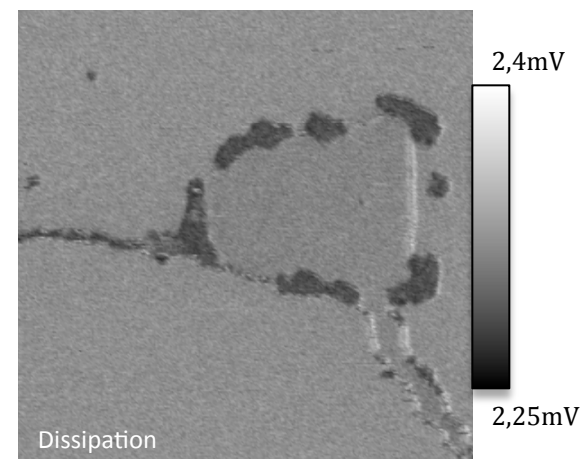
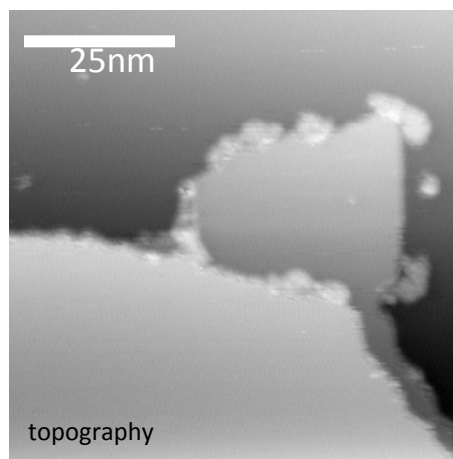
U < 1,5kV

U ≥ 3kV

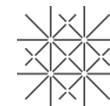
Lu53 / Cu(111)



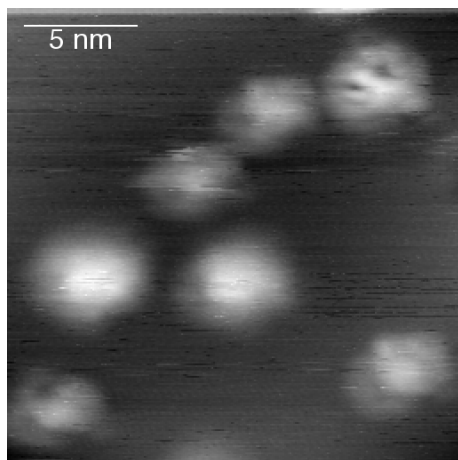
Toluene: isopropanol (2:1).



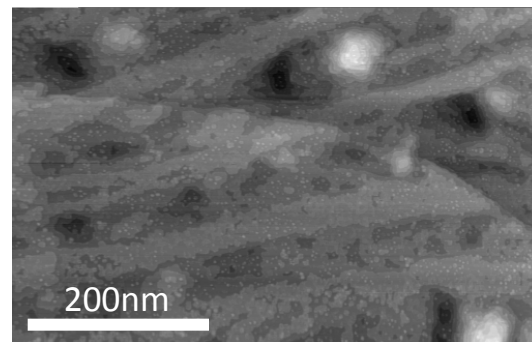
ESD of NanoDiamond



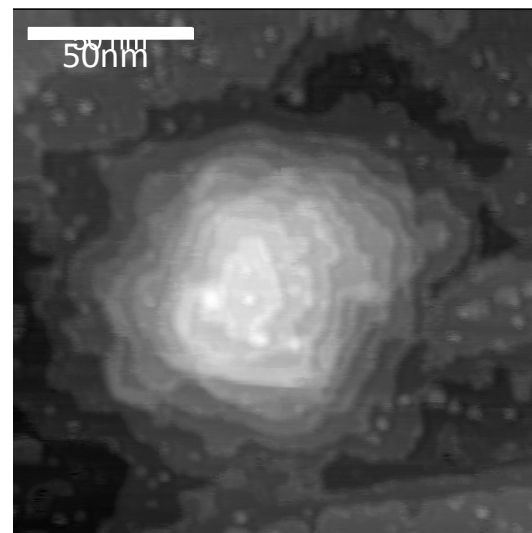
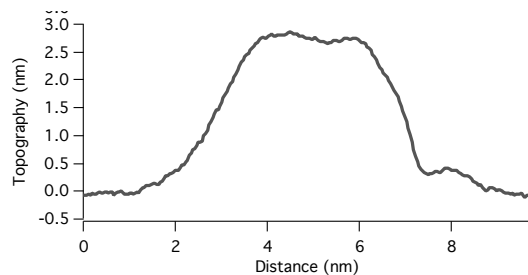
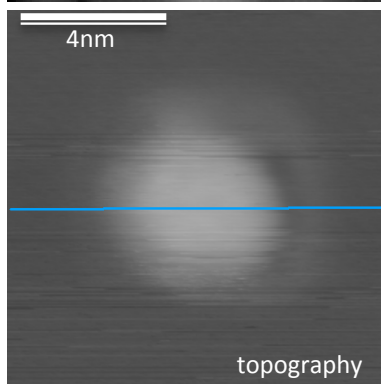
Nanodiamants synthétisés à l'institut de saint louis.
2 solutions aqueuses: 2-5nm et 50nm de diamètre.



ND 50nm/KBr(001) :



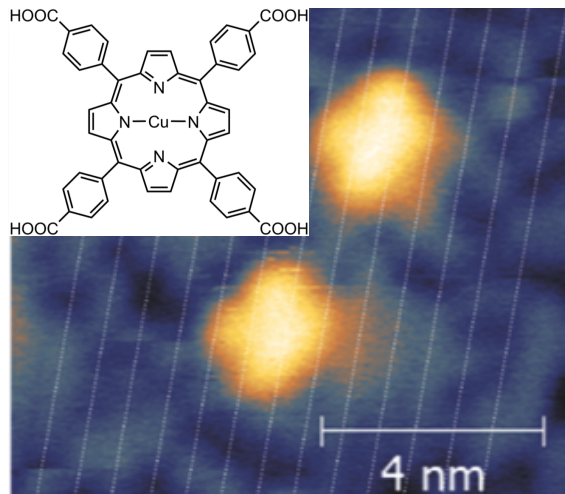
ND 2-5nm/KBr(001)



Conclusion

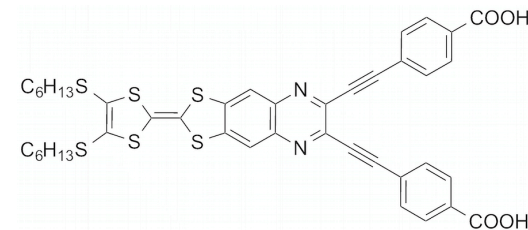


- ❖ Dépôt de molécules sur isolants massifs possible par ESI.
- ❖ Différents taux de couverture possible.
- molécules isolées.



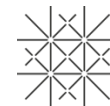
Cu-porphyrin/TiO₂ rutile (110)

R. Jöhr et al. *The Journal of Chemical Physics*, 143, 9 (2015)

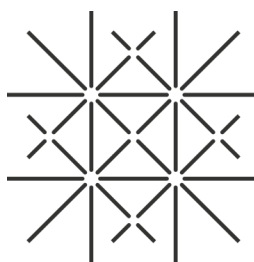
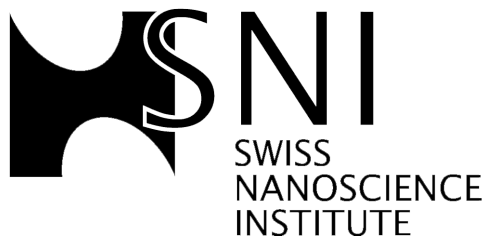


ESI : TTF-Dye
TiO₂ rutile ou anatase

Acknowledgements



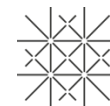
Vielen Dank für Ihre Aufmerksamkeit !!



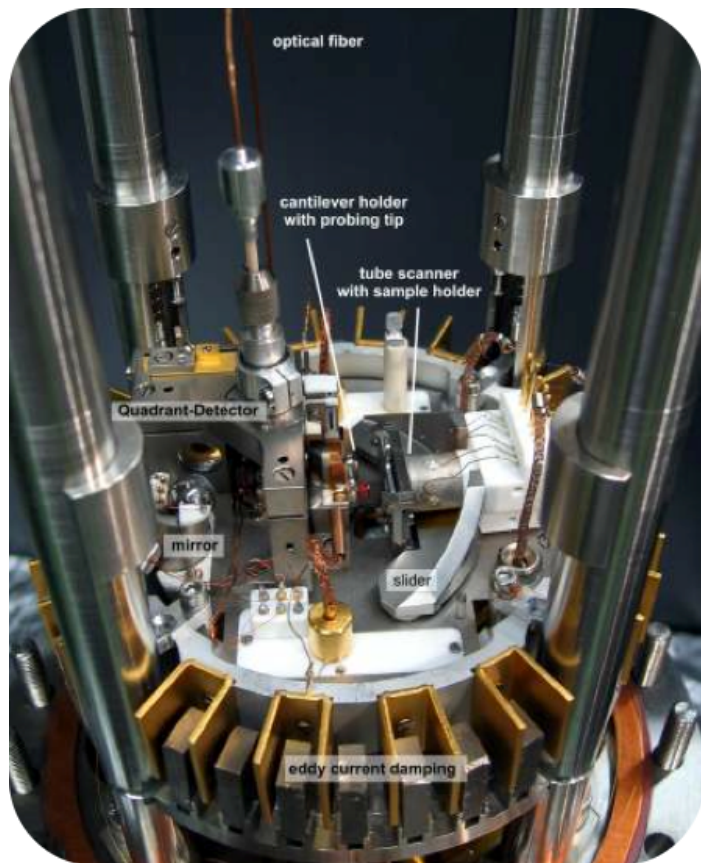
UNI
BASEL



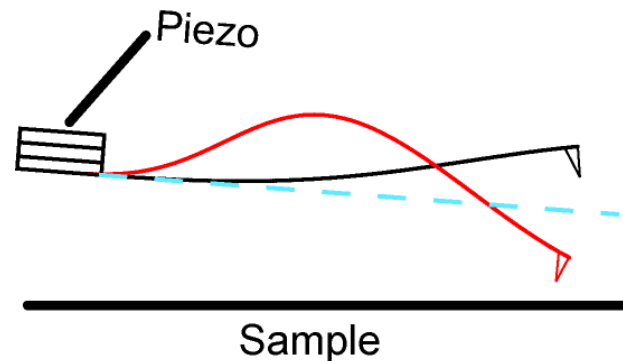
Dispositif et caractérisation



UHV home built RT AFM



ncAFM bimodale



Cantilever Si : PPP-NCL, chauffage 1 h @ 100°C, Sputtering 2min.

$f_{1st}=150kHz$	$f_{2nd}=1MHz$	$f_{TR}=1.5MHz$
$Q_{1st}=30000$	$Q_{2nd}=10000$	$Q_{TR}=100000$

$k_{2nd} \gg k_{1st}$
S. Kawai, et al., APL 86 193107 (2005)